First Exam	Name (PRINT)	Loot First
		Last, First
Chemistry 3332	Signature	
February 17, 2006	ID#	
	<u></u>	

Please circle class time.

Dr. Bean's 10:00 AM

Dr. Bean's 1:00 PM

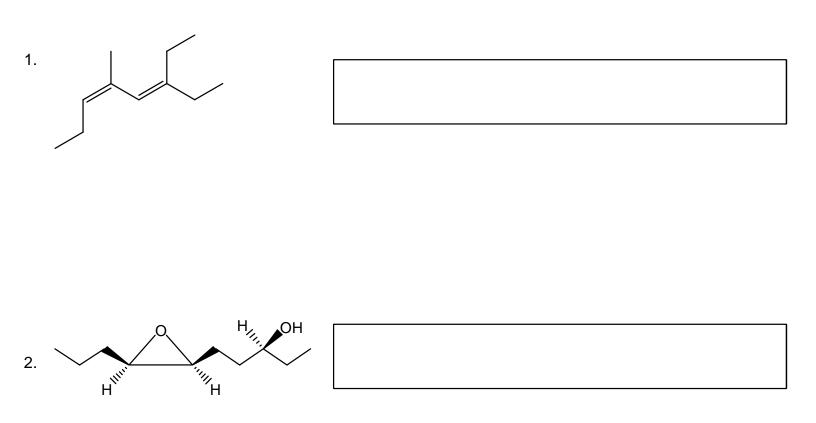
Page #	Score	
1. 12 pts.		
2. 20 pts.		
3. 18 pts.		
4. 18 pts.		
5. 11 pts.		
6. 11 pts.		
7. 10 pts.		

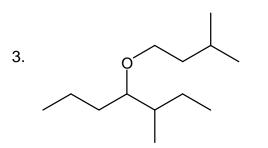
TOTAL_____

Note: Present your student ID when you return the exam booklet

A. Nomenclature: (12 points, 4 points each)

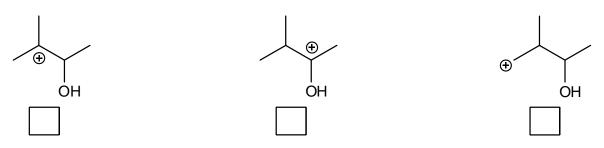
Give an acceptable name for each of the following compounds. Be sure to note stereochemistry where appropriate.



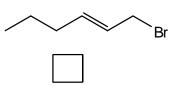


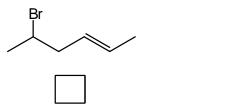
B. Facts: Total Points = 20

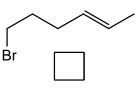
1. Place the following carbocations in order of increasing stability. (1=least stable, 3=most stable) (3 pts.)



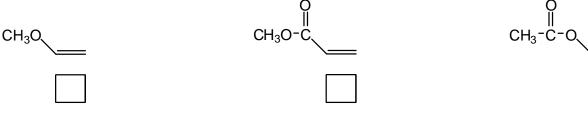
2. Place the following compounds in order of increasing reactivity in an $S_N 2$ process. (1=least reactive, 3=most reactive) (3 pts.)



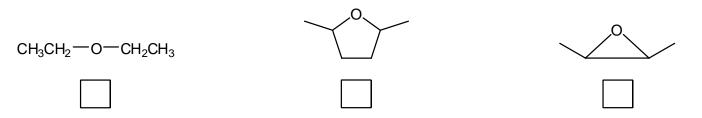




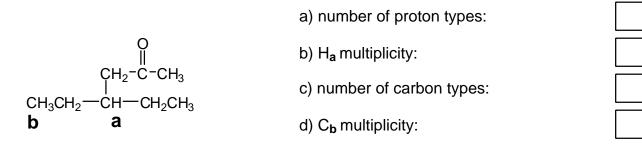
3. Place the following molecules in order of increasing reactivity in a Diels-Alder reaction. (1=least reactive, 3=most reactive) (3 pts.)



4. Place the following compounds in order of increasing reactivity with HI. (1=least reactive, 3=most reactive) (3 pts.)

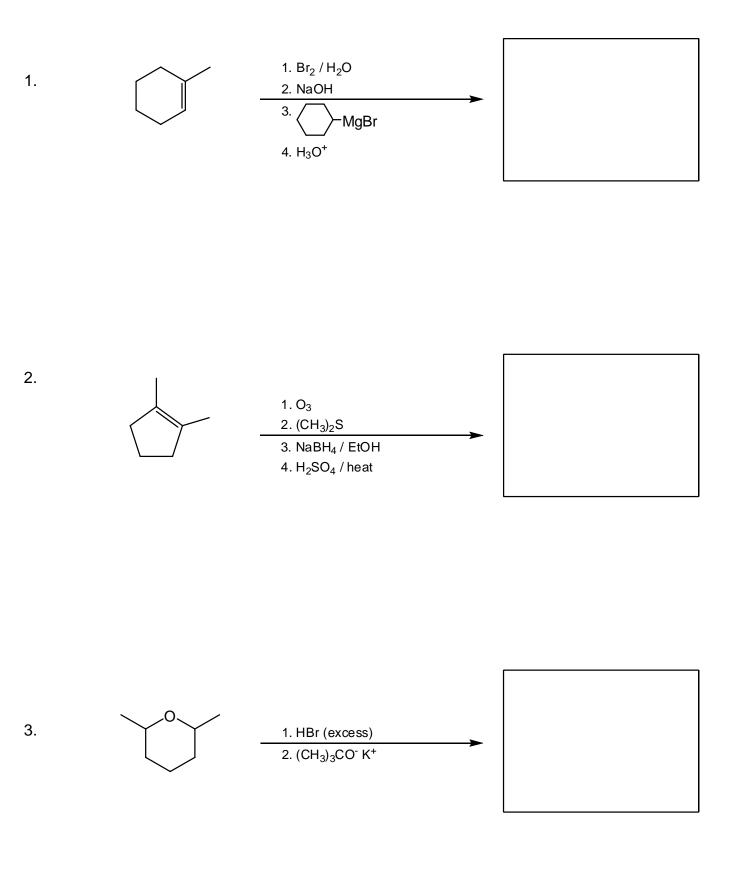


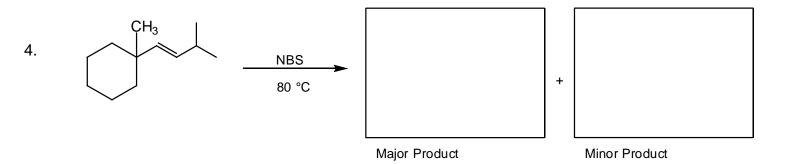
5. Answer the following questions about the compound below. Place the answers in the appropriately labeled boxes. a) How many distinct types of protons does the compound have? b) What is the theoretically predicted multiplicity (splitting pattern) of the signal for the proton labeled **a**? c) How many distinct types of carbons does the compound have? d) What is the multiplicity of the signal in the proton coupled ¹³C NMR spectrum for the carbon labeled **b**? (8 pts.)

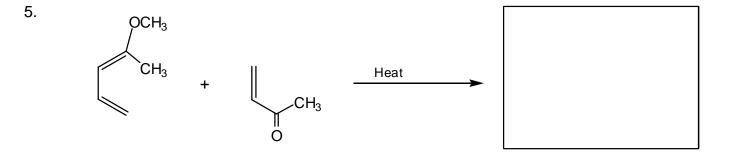


C. Reactions: Total = 36 points, 6 points each

Please provide the major product (unless otherwise indicated) in the answer box. Be sure your drawing indicates stereochemistry if applicable. Partial credit is awarded only when intermediate products are shown below the reaction.







6.

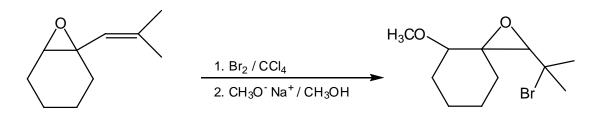
$$CH_{3}-C-H$$

$$(H_{3}-C)=C O Na^{O}$$

$$(H_{3}-$$

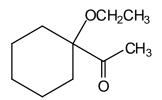
D. Mechanisms: (11 points)

Please provide a clear mechanism for the reaction below. Use curved arrow notation to indicate "electron flow". Show all intermediates and all formal charges.



E. Synthesis: (11 points)

Synthesize the molecule below using any of the following reagents: cyclohexane and alkanes, alkenes, alkynes or alcohols of **two carbons or less**; any oxidizing or reducing agents, and any peroxyacids.



F. Spectroscopy (10 points)

A compound with the formula C_9H_8O exhibits the IR, ¹H NMR and proton decoupled ¹³C NMR spectra shown below. Please identify this compound and draw the structure in the box provided below.

