

EXAM 3, Sp '08

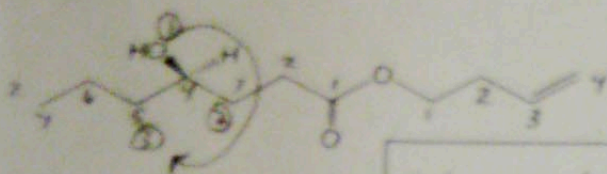
8. Nomenclature: (15 points)

Give an appropriate IUPAC name for each of the following compounds. Be sure to indicate the stereochemistry where appropriate.



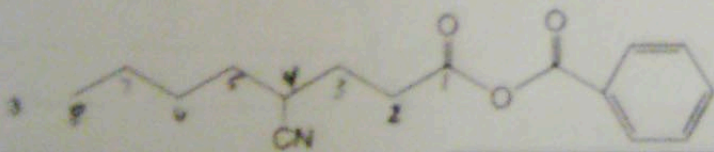
(E)-3-methyl-N-phenyl-2-hexenamide

1 1 1 1 1
2R
(2E)



3-butenyl (R)-4-hydroxyheptanoate

1 1 1 1 1



benzoic 4-cyanoctanoic anhydride

1 1 1 2

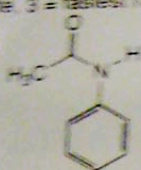
9. Facts: 12 points

Rank the following compounds in order of increasing reactivity with H_2O (1 = slowest

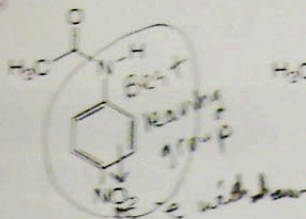
3 = fastest) (3 pts)

B. Facts: 12 points

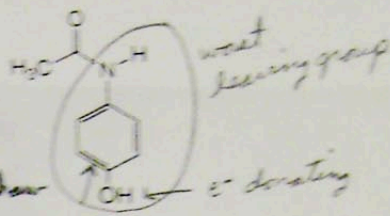
1. Rank the following compounds in order of increasing reactivity with H_2O . (1 = slowest rate, 3 = fastest rate) (3 pts)



2

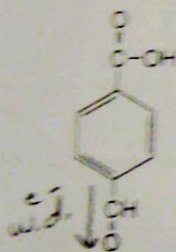


3

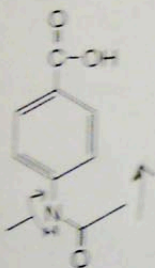


1

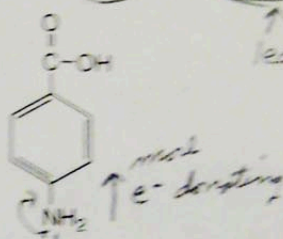
2. Rank the following compounds in order of increasing pKa. (1=lowest pKa, 3=highest pKa) (3 pts)



1

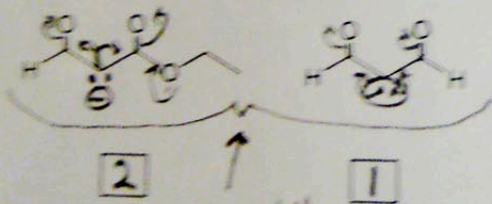


2

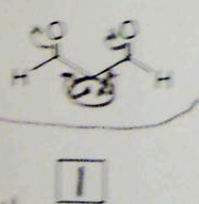


3

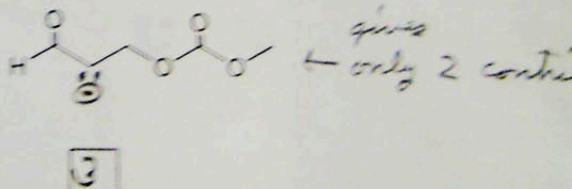
3. Rank the following compounds in order of increasing pKa. (1=lowest pKa, 3=highest pKa) (3 pts)



2

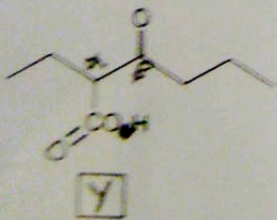


1

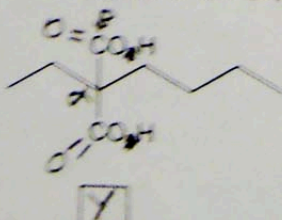


3

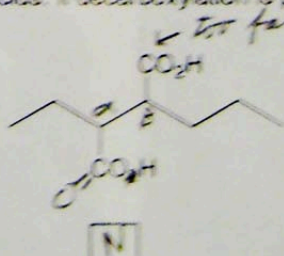
4. Consider the decarboxylation at $180^\circ C$ of the following carboxylic acids. If decarboxylation is possible, place **Y** (for yes) in the box. If not, place **N** (for no) in the box. (3 pts)



Y



Y

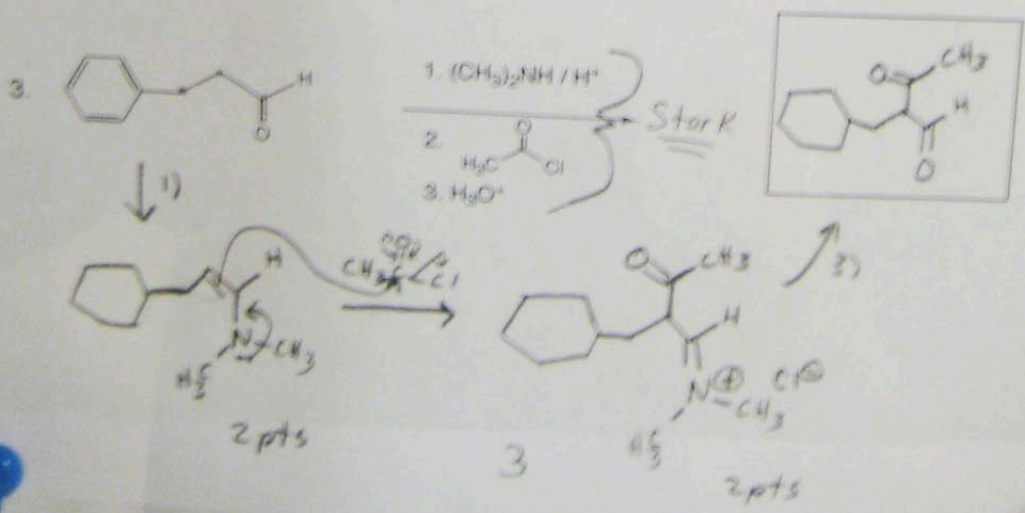
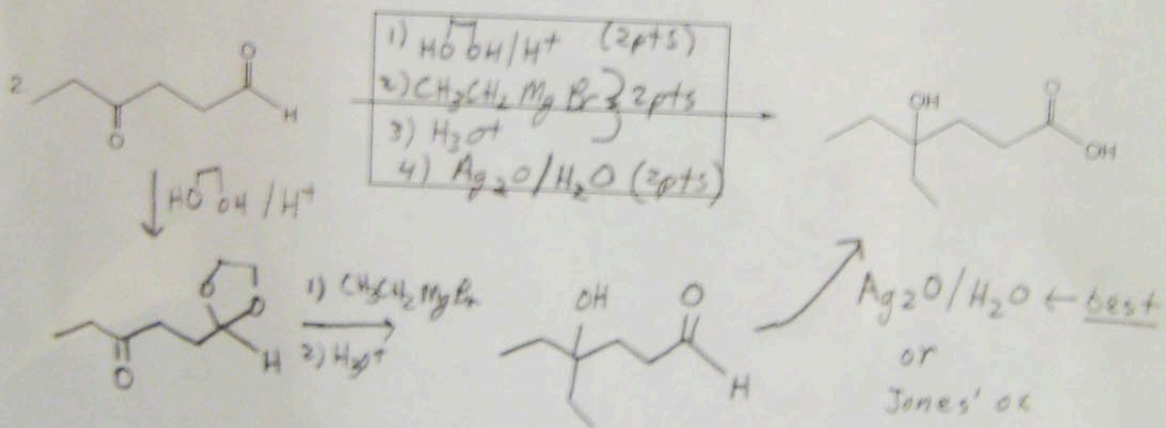
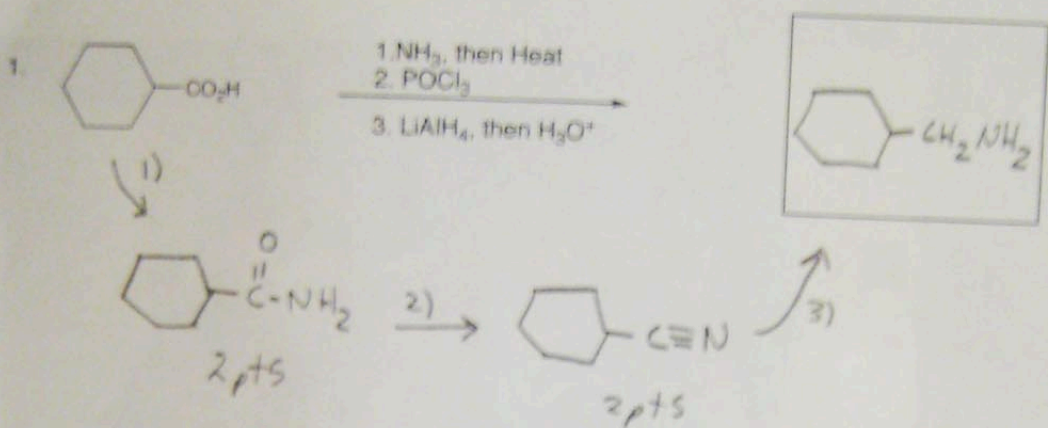


N

β -Keto acids necessary

C. Reactions: Total = 36 points, 6 points each

Please provide the starting material, reagents or major product in the answer box. Be sure your drawing indicates **stereochemistry** if applicable. Partial credit is awarded only when intermediate products in a multi-step reaction are shown below the reaction.



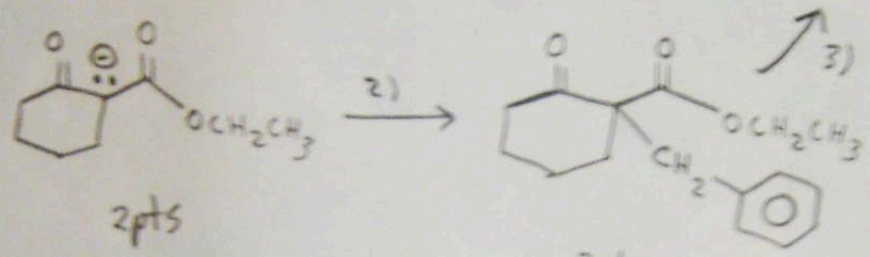
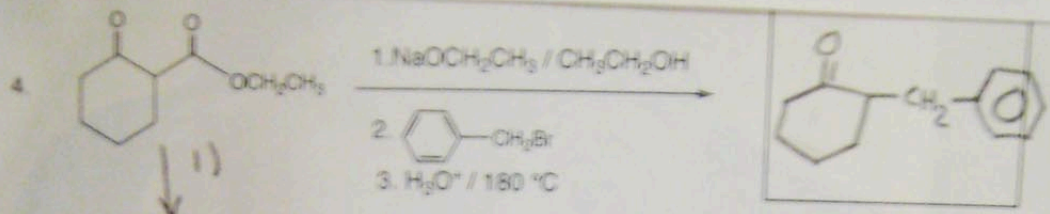
H₃⁺

2 pts

3

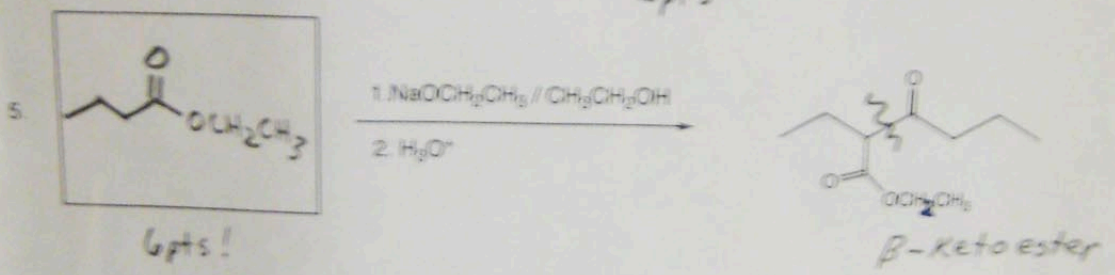
H₃⁺ N⁺-CH₃

2 pts

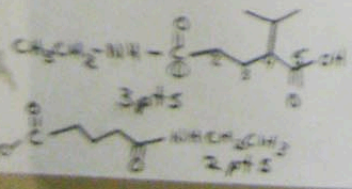
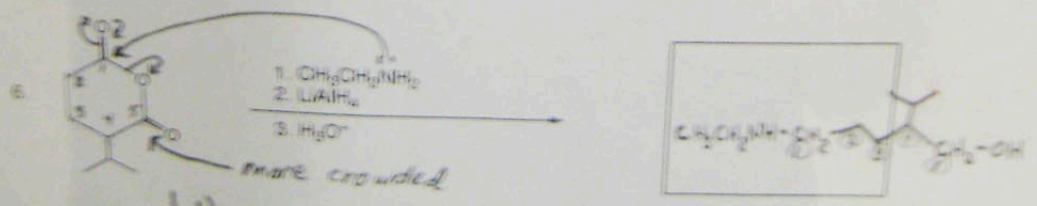


2 pts

2 pts



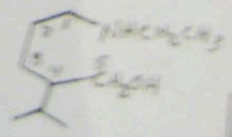
6 pts!



3 pts

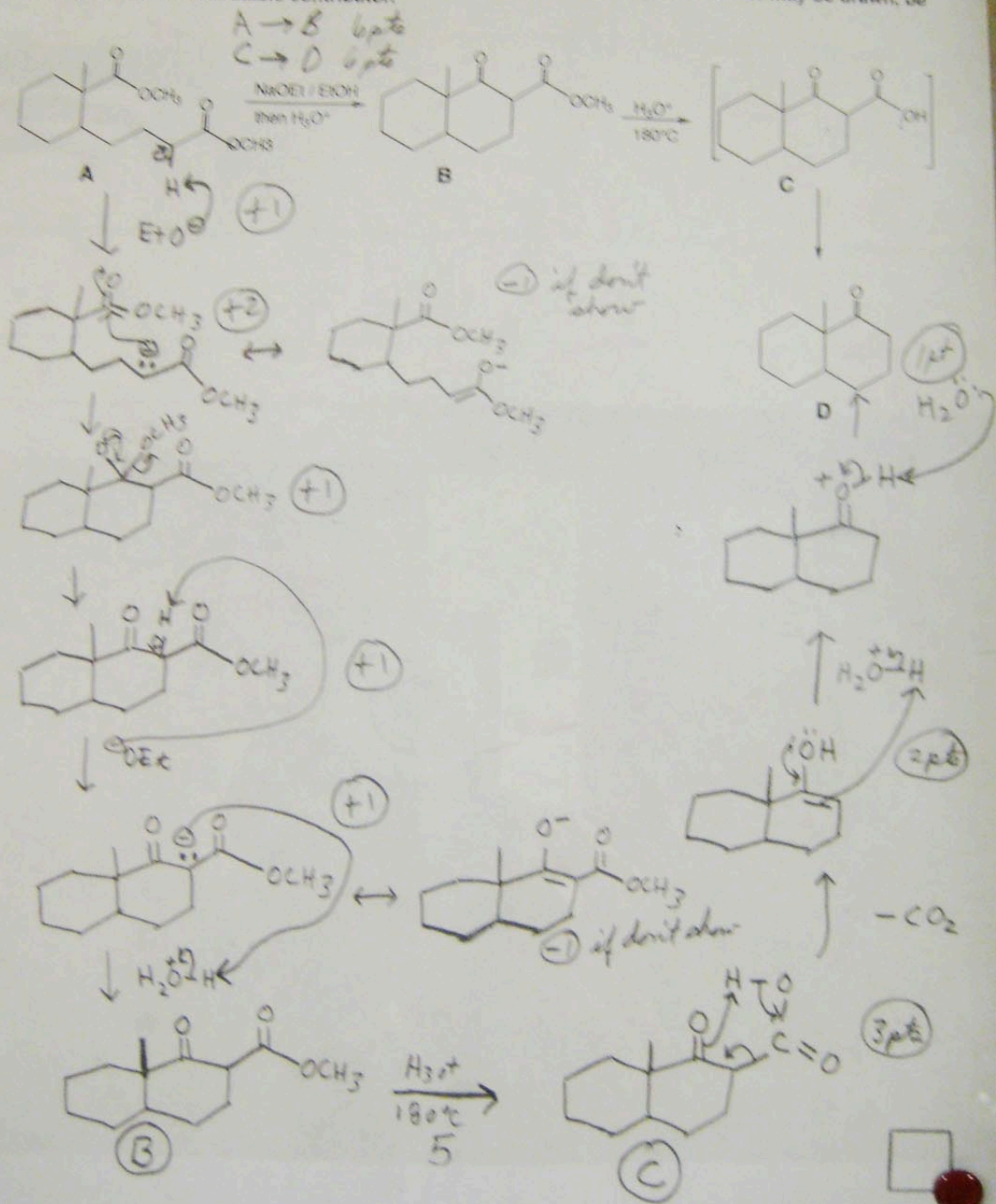
2 pts

4



D. Mechanism (12 points)

Provide a clear mechanism to explain the formation of the series of products below. Do not show the transformation of B to C. Remember to show only one step at a time (NO SHORTCUTS!). Show all intermediates and all formal charges. When more than one resonance contributor may be drawn, be sure to draw the most stable contributor.

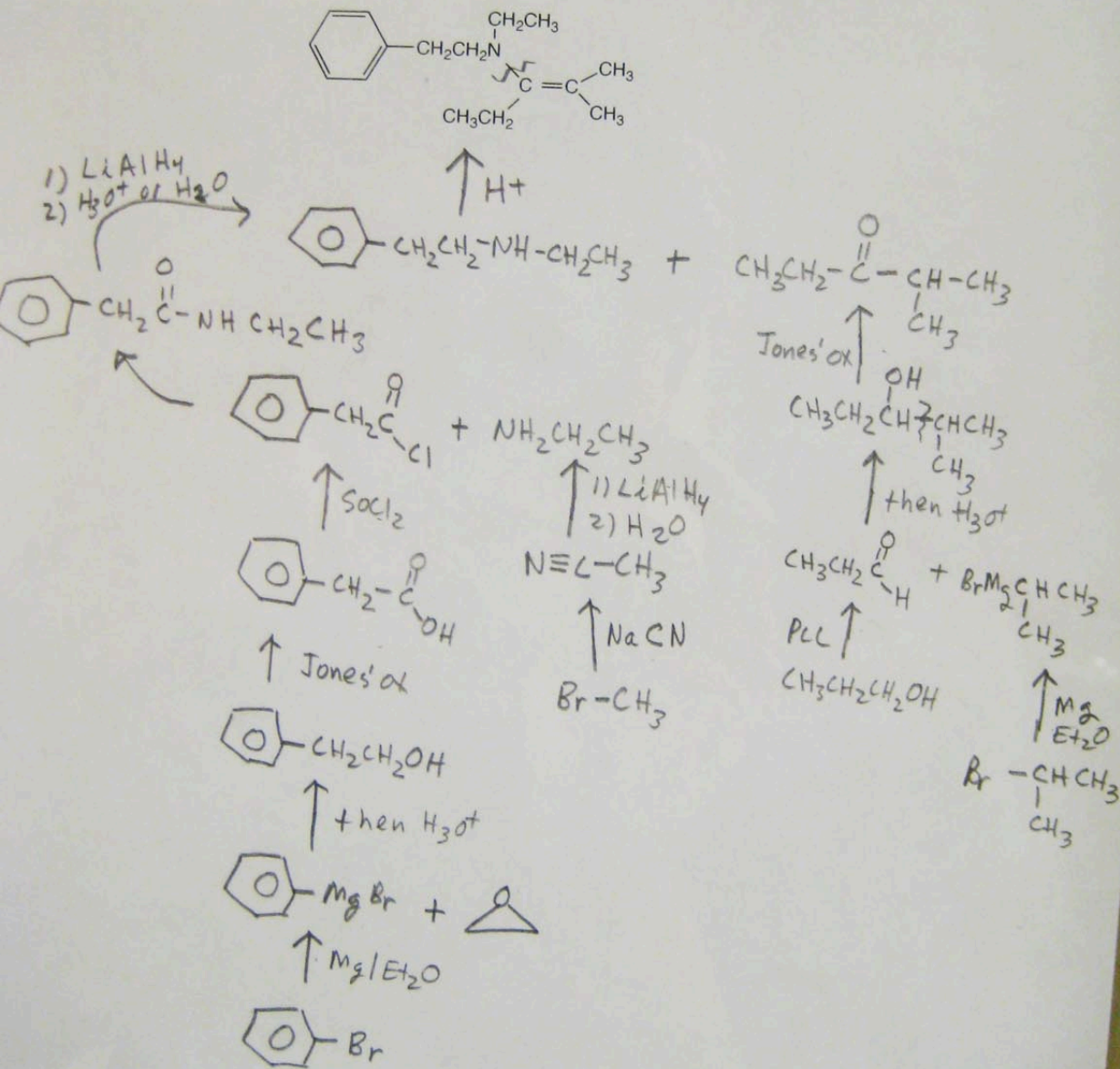


E. Synthesis: 13 Points

Synthesize the molecule below using any of the following reagents: benzene, bromobenzene, any alkenes, alcohols, or alkyl halides of three carbons or less; ethylene oxide; any inorganic reagents, oxidizing or

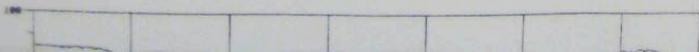
E. Synthesis: 13 Points

Synthesize the molecule below using any of the following reagents: benzene, bromobenzene; any alkenes, alcohols, or alkyl halides of **three carbons** or less; ethylene oxide; any inorganic reagents, oxidizing or reducing agents, and any peroxyacids.



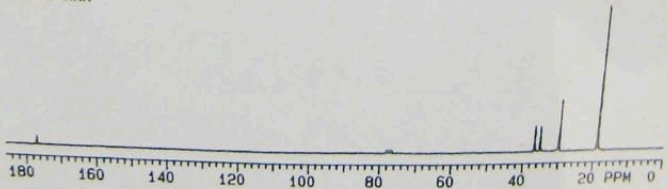
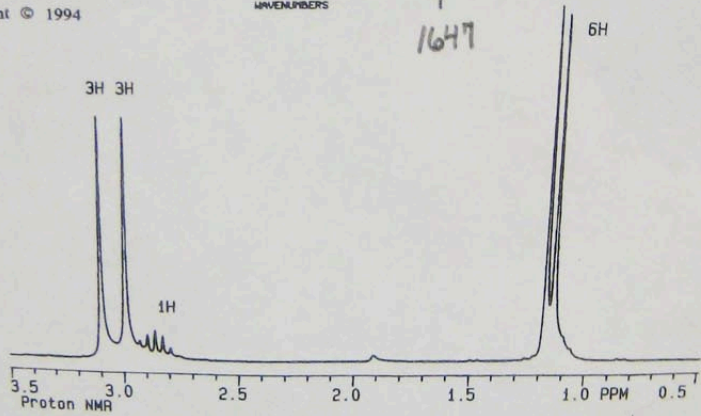
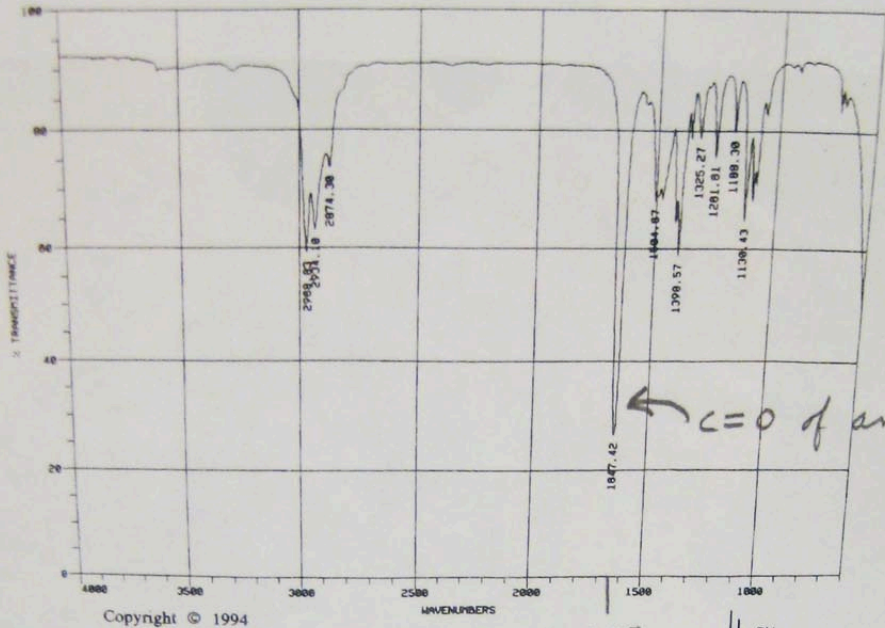
F. Spectroscopy: 12 Points

A compound with the formula C₆H₁₃NO exhibits the IR, ¹H NMR and proton decoupled ¹³C NMR spectra shown below. Please identify this compound and draw the structure in the box provided below.



F. Spectroscopy: 12 Points

A compound with the formula $C_6H_{13}NO$ exhibits the IR, 1H NMR and proton decoupled ^{13}C NMR spectra shown below. Please identify this compound and draw the structure in the box provided below.



partial credit:

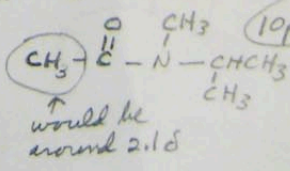
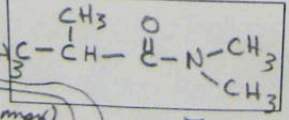
amide: 3pts

isolated CH_3 : 1pt each (2 max)

isopropyl: 3pts

1H adj to meth...

CH_3 adj. to 1H: 1pt each (2 max)



ONLY if NO isopropyl

