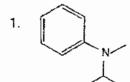
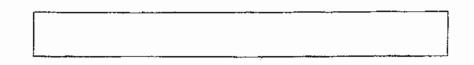
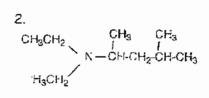
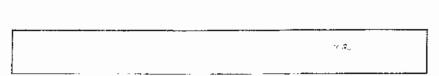
## A. Nomenclature: Total = 12 points, 4 points each

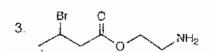
Give an acceptable name for each of the compounds below.











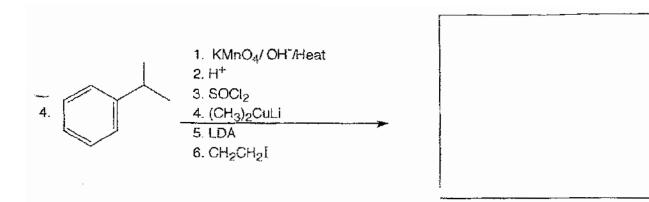


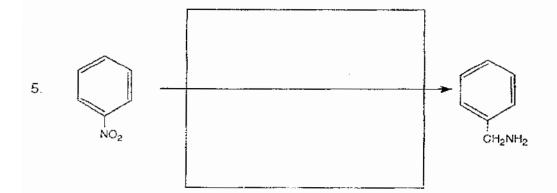
## D. Synthesis: (17 pts)

Synthesize the molecule below using any of the following reagents: benzene, cyclohexane, alcohols, alkanes, alkenes, and/or alkynes of two carbons or less, any inorganic reagents, any oxidizing or reducing agents, and any peroxyacids.

## C. Mechanism: (17 pts)

Provide clear mechanisms for the following reaction. Use curved arrow notation to indicate electron flow. Show all intermediates and all formal charges. If there is more than one resonance structure, you must show the "best" (i.e., lowest energy) structure.





6. 
$$H_2C = CH - C = N + CH_3O CCH_2COCH_3 = \frac{1. NaOCH_3/CH_3OH}{2. H_3O^+ (xs)/Heat}$$

F, 01 3

